

Les Rosiers



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(Avery Daugherty)

LES ROSIERS



RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 52 pages. OCLC Number: 53057092 Excerpt: . . . memorandum from EPA Administrator, Carol Browner, dated March 21, 1995, Subject: Guidance on Risk Characterization. The initial literature search strategy employed for this compound was based on the CASRN and at least one common name. However, because the initial search for acetone yielded more than 1, 500 hits, subsequent searches limited to review articles (all years) and for acetone as a word in the title (1993-present) were conducted. The following data bases were searched: TOXLINE (all subfiles), MEDLINE, CANCERLIT, TOXNET HSDB, IRIS, CCRIS, EMIC (1991-present), and GENE-TOX , and RTECS. 2. CHEMICAL AND PHYSICAL INFORMATION RELEVANT TO ASSESSMENTS Chemical and physical properties of acetone are listed in Table 1. Acetone is used primarily as a chemical intermediate and as a solvent in both chemical and pharmaceutical applications. Acetone is also an endogenous compound utilized in intermediary metabolism (ATSDR, 1994; WHO, 1998). As the data below indicate, acetone is completely miscible with water and highly volatile. Table 1. Physicochemical Data for Acetone Parameter Value Reference Synonyms 2-propanone; dimethyl ketone Budavari et al. , 1996 CAS registry no. 67-64-1 Chemical formula CH₃COCH₃ Budavari et al. , 1996 3 3 Molecular weight 58. 08 Budavari et al. , 1996 Physical state liquid Budavari et al. , 1996 Vapor pressure at 20 C 181. 72 mmHg ATSDR, 1994; WHO, 1998 Melting point-94 C Budavari et al. , 1996 Boiling point 56. 5 C Budavari et al. , 1996 Solubility in water miscible Budavari et al. , 1996 Log K-0. 24 ATSDR, 1994; WHO, 1998 OW-5 3 Henrys Law Constant 4. 26 10 atm-m mol ATSDR, 1994; WHO, 1998 3 Conversion factors in...



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